ASTRO 3D

Understanding Galaxies Using Coloured Flames

SAFETY WARNING: Wear protective gloves and glasses and work in a well-ventilated room. Observe safety rules when working with fire and flammable liquids.

Equipment

- Tea-light candle holders
- Methylated spirits, isopropyl alcohol, or similar lighter fluid
- Long-handled lighter
- Cooking tray to smother flames
- The following ingredients:



| Chemical | Common Item | Colour |
|--------------------|--|----------|
| Alcohol/methanol | Methylated spirits, isopropyl alcohol | Blue |
| Boric acid | Borax (multipurpose cleaner) | Green |
| Sodium chloride | Table salt | Yellow |
| Calcium chloride | Damp Rid (moisture absorber) | Orange |
| Lithium | Lithium batteries (see extraction steps) | Pink/red |
| Potassium chloride | Replacement table salt (KCl) | Purple |
| Magnesium | Epsom salts | White |

Method

- 1. Place a large spoonful of the chemical into a tea light candle holder. Carefully pour in the lighter fluid to cover the chemical completely.
- 2. Use the long-handled lighter to set the fluid alight.
- 3. Watch as the colour of the flame changes.
- 4. Place the metal tray onto the candle holder to smother the flames.
- 5. Carefully discard of products.

Notes

- If the chemical is in a powder or liquid form, it will improve the colour. Chemicals that come in a flake form can be crushed into a finer powder.
- Let the chemical sit in the lighter fluid for a while before burning to improve the colour.
- A lighter fluid that burns with a clearer flame will produce better results.
- If you cannot extract lithium from a battery, it is also possible to use strontium chloride to make a red flame.
- This experiment also works well in a science classroom, using the salts in a spray test over a Bunsen burner.

DISCLAIMER: ASTRO 3D excludes all liability to any person arising directly or indirectly from undertaking this activity.

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Extracting lithium from battery

SAFETY WARNING: Wear protective gloves and glasses as the materials inside the battery can burn exposed skin. It is very easy to short-circuit a battery, causing it to catch fire. If this happens, simply place the battery on a fire-safe surface and let it burn itself out.

- Use wire cutters to cut the casing around the positive end of an unused standard lithium battery. At this stage, beware of the battery short-circuiting it will begin to feel warm. Cutting the top away as quickly as possible and removing the upper terminals will minimise the risk of a short circuit.
- The lithium is rolled up as a foil inside the battery (it is the dull, softer foil). Remove it carefully, cut a piece of lithium foil and place straight into lighter fluid. The lithium will react with oxygen in the air and so needs to be placed straight into the lighter fluid.
- The remaining battery components should be wrapped up and discarded.